

O71

SELECTIVE INCORPORATION OF F,CF₃ AND CF₃S GROUPS INTO AROMATIC SUBSTRATES

A. Beaumont, J. H. Clark, J. Denness, C. Jones, M. McClinton and A. Wynd
Chemistry Department, University of York, York YO1 5DD (U.K.)

The use of modern inorganic reagents such as Bu₄NF, Ph₄PHF₂, 'CuCF₃' and alumina-CuSCF₃ enables the preparation of a wide range of aromatic molecules containing F,CF₃ and CF₃S groups [1-3]. Fluorodenitration and fluorodechlorination can be used for the selective fluorination of aromatic substrates. The susceptibility of the nitro groups to exchange by fluorine is influenced by both steric and electronic factors and the relative rates of fluorodenitration and fluorodechlorination can be influenced by the reaction conditions. Burton's reagent can be used to selectively replace aromatic chlorines ortho to chelating groups by CF₃. The novel reagent alumina-CuSCF₃ is a convenient and easily prepared source of SCF₃ for aromatic substitution. Some of the subsequent reaction chemistry of molecules containing F,CF₃ and CF₃S groups will be described.

- 1 J.H. Clark, M.A. McClinton, C.W. Jones, P. Landon, D. Bishop and R.J. Blade, Tetrahedron Lett., in press.
- 2 J.H. Clark, M.A. McClinton and R.J. Blade, J. Chem. Soc. Chem. Commun., 638, (1988).
- 3 J.H. Clark and D.J. Macquarrie, Tetrahedron Lett., 111, (1987).